Study of H⁻ extraction from a single-hole plasma electrode of C12A7 electride: A way to a Cs-free H⁻ Source

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What is a C12A7 Electride

H. Hosono and his coworkers have found that 12CaO.7Al2O3 (C12A7) can be transformed to an electride , which was formed by removal of clathrated oxygen ions from the cages in a single crystal of C12A7 leading to formations of high-density electrons in the cages. The connected cages form a new conduction band called "cage conduction band" (CCB), below the cage flame conduction band minimum (FCBM). The work function (WF) is 2.4 eV.







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Experimental observation I - atomic hydrogen injection

Doshisha



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H⁻ spectra measurement

Aix Marseille University



measured by Hiden EPQ300

• RF plasma $T_e \sim 3 \text{ eV}$, $n_e \sim 10^8/\text{cm}^3$ without B

- μ W plasma T_e ~ 1 eV, n_e ~ 10⁹/cm³ •
- H_{3}^{+} ions are dominated.
- HIDEN particle analyzer
- L probe
- Heater (<650°C) & TC



H-/H+ ion spectra from the sample surface were

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H⁻ spectra measurement

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Experimental observation II

Marseille

Electride





Experimental observation II

10^{5} 2 Pa, D₂ RF plasma, 250 W 10^{4} Count at Peak 1000 回目) ELECTRIDE(2回目) – Mo 100 40 80 120 0 -V(target)

Marseille

Details of experimental results will be presented on Friday, by

Roba Moussaoui Aix-Marseille Univ..

Id: 37 Negative-ion production study on nanoporous 12CaO.7Al2O3 electride surface in low pressure H2 plasma

Experimental observation II



Using the pseudo exponential work-function (ϕ) dependence, the ratio of H⁻yield from a cesiated molybdenum surface of the lowest work-function to that from a clean molybdenum surface (ϕ =4.3 eV) can be evaluated to be about 40 at 100 V, while the ratio of total H⁻ yield from a C12A7 electride surface bombarded at 80 V to that from a clean molybdenum surface is about 10. The ratio became larger as the bias voltage V_s became lower, and the ratio was 50 at V_s = 20 V. **Rapid report : Applied Physics Express 11, 066201 (2018)**

Study of H- extraction from a single-hole plasma electrode of C12A7 electride.

OUTLINES

1. Objectives : Aiming a Cs-free H⁻ Source

Can a high production rate be expected in a real ion source when the plasma grid is fabricated with the C12A7 electride?

- 2. Experimental Setup
- 3. Experimental Results
- 4. Discussions
- 5. Future Plans



Study of H- extraction from a single-hole plasma electrode of C12A7 electride.

Experimental Setup

- 1. A small ECR plasma source of 50W(max) (Tamaoki Electronics Co. LTD)
- 2. An ECR plasma is diffused to a plasma chamber of ~ 10 cm³ toward plasma grid.
- 3. A plasma grid of C12A7 electride or clean Mo of 16 mm φ, 2 mm t was used. The extraction hole is a 2 mm φ tapered.
- 4. Pretreatment of C12A7 electride =>annealing at ~520°C for 2 hrs.
 => take-out to air (dry) after cooling and install to the ion source.
- **5.** A single stage extraction grid of 6 mm ϕ is used.
- 6. A Faraday cup with permanent magnets is installed, and the current was measured by a picoammeter, ADVANTEST TR8641.
- 7. He-Cd Laser (Kimmon Electric Co. LTD) of 13 mW, 325 nm or another diode laser is injected.

Experimental Setup



Experimental Results : FC current



- **Higher production rate** was observed with a C12A7 Electride PG, than a clean Mo PG, by a factor of 40 - 100.
- **Effect of electron** suppression magnets near the extraction region is clearly seen with a Mo PG.

Experimental Results : FC current



FC Current with C12A7 Electride PG, increases as $(V_{acc})^{1.2}$, and that with Mo PG as $(V_{acc})^{0.9}$.

More data will be presented by M. Kobayashi in the PS #2, Tuesday

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Experimental Results : Extraction



Acceleration current was compared with the FC current.

Remind:

FC current would be less than the total H⁻ current extracted, because the extraction was not optically optimized.

Acceleration current contained secondary electron current.

Nevertheless, the ratio of H⁻ /e in the extraction current might be larger than 1/23 for C12A7 electride PG, and 1/1300 for Mo.

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Discussions and Summary

- C12A7 Electride is a stable, robust, machinable material, with low work function, and fabrication of PG for a H⁻ source is possible.
- Preliminary study of H⁻ extraction from C12A7 electride PG has been carried out using a compact ECR ion source.
- The extracted H⁻ current was compared with that from a clean Mo PG, and the former is higher than the latter by factor 40 - 100.
- Margins

Extraction voltage should be increased, up to ~ several kV. Plasma density, not measured in this work, could be increased. Surface condition of C12A7 electride was not optimized. The surface conductivity was not high enough. Work function was not low enough.

 Nevertheless, the ratio of H⁻ from C12A7 electride to that from clean Mo is consistent with our results of Marseille experiment*.

* Roba Moussaoui et al.[Id: 37], will be presented on Friday morning. 2018/9/3 M.SASAO@Novosibirsk_NIBS2018

Future Plans

- Present compact ECR ion source of C12A7 elctride (Source 1) will be upgraded, so that we can anneal the PG in-situ, measure the ϕ , and measure the plasma parameters.
- Source 2 a bucket type source having an ECR plasma source and a three electrode extractor: acceleration and deceleration grids, with the electride plasma grid having an indirectly heating system, is now under construction.*





Extraction Grid and Acc.Dcc Grid.

* M. Kobayashi et al.[Id: 52], in the PS #2, on Tuesday.

supplement



Measurement of Photoelectric Response

The time evolution of the target temperature and photoelectric current for 325 nm, 405 nm, 460 nm, 532 nm diode laser injection on to the C12A7-Electride. The present data do not fit into the

Fowler's formula,

$$\frac{I(\chi_0 - h\nu)^{1/2}}{T^2} = Af(\mu) = Af\left(\frac{h\nu - \chi}{kT}\right)$$
$$f(\mu) = \frac{\pi^2}{6} + \frac{1}{2}\mu^2 - e^{-\mu} + \cdots$$

The fitting photoelectric currents in the region of high energy photon incidence to this formula indicated that the work function at 450° C is lower than 2.7 eV.



Experimental observation II



Experimental observation II

Marseille **Ar Bombardment** Vs= -60 V - Argon plasma Hydrogen plasma 0.1 Normalized NIEDF **Comparison of normalized Negative** 0.01 **Ion energy Spectra** 1E-3 bombarded in Ar plasma and H2 -10 20 30 -20 10 40 50 plasmas at Vs = -60 V Energy (eV)

Figure 63 comparison of normalized NIEDF measurement in Ar plasma and in H2 plasma at VS= -60 V

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Experimental observation II

Time evolution of Negative ion yield after 10 min. of high energy bombardment followed by a heating in vacuum and not followed by a heating in vacuum



Dominant process is desorption by sputtering

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Future plans

- Compact H- source with a plasma grid of C12A7 Electride is now under construction.
- Absolute H- current from the source will be measured.

Thank you for your attention





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H⁻ ion

Flectro

C12A7:e⁻⁻

Kitano et al., NATURE CHEMISTRY | VOL 4 | NOVEMBER 934 2012



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Experimental observation II - H⁻ spectra measurement

Marseille





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Thermocoax heater

Plasma Surface Interaction – Refection & Desorption





Experimentally measured energy spectra for different tilt angles of the HOPG sample

Refection : Mirror/Cone Angular Distribution, continuous up to the incident energy distribution

Desorption by Sputtering : Normal Angular Distribution, low energy distribution.

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FIG. 2. Dependence of the H⁻ yield upon the work function of the con-

Dependence of H⁻⁻ production upon the work function of a Mo surface in a cesiated hydrogen discharge

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(Received 23 October 1989; accepted for publication 30 January 1990)

The photoelectric work function of a cesiated molybdenum converter surface in a cesiated hydrogen discharge and the negative-hydrogen-ion (H^-) current produced at the surface were simultaneously measured. With the negative bias potential of the converter constant, the H^- yield increased exponentially as the work function was decreased by introducing Cs into the discharge. The yield of H^- current was always higher for a higher bias potential on the surface, provided the measured surface work function was nearly the same. When the concentration of Cs in a discharge was nearly constant, the bias potential of the surface at which the H^- production became maximum was observed. At this bias potential, the surface work function was close to the work-function minimum.

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